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Resolution of (4RS,5RS)-4,5-diphenylimidazolidine-2-thione using pentafluorophenyl active esters

Anna Andreou, Najla Al Shaye, Hannah Brown, Jason Eames*

Department of Chemistry, University of Hull, Cottingham Road, Kingston upon Hull HU6 7RX, UK

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Article history: Received 30 July 2010 Revised 15 October 2010 Accepted 29 October 2010 Available online 4 November 2010 ABSTRACT

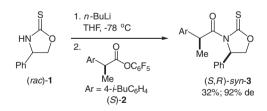
(4RS,5RS)-4,5-Diphenylimidazolidine-2-thione is resolved efficiently by treatment with NaHMDS and an enantiomerically pure pentafluorophenyl active ester. The levels of diastereocontrol were excellent (up to 90% de).

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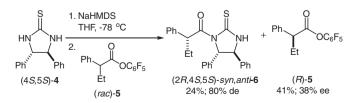
The separation of enantiomeric substrates¹ using parallel resolution is becoming an increasingly popular method.² Over the last few years, we have been interested in the kinetic resolution of oxazolidine-2-ones³ and oxazolidine-2-thiones,⁴ such as 4-phenyl-oxazolidine-2-thione (*rac*)-1, using an enantiomerically pure ibuprofen-derived active ester, 2-(4-isobutylphenyl)propanoate (*S*)-2, to give the corresponding oxazolidine-2-thione adduct (*S*,*R*)-*syn*-3 in 32% yield with 92% de (Scheme 1).³ The levels of diastereocontrol were found to be excellent for a wide range of structurally related profen-based active esters. This methodology has more recently been applied to the resolution of racemic alcohols, such as 1-phenylethanol,⁵ and carboxylic acids.⁶

Using our standard protocol,³ we were interested in studying the resolution of C₂-symmetric 4,5-diphenylimidazolidine-2thione (4RS,5RS)-4, as this substrate contained all the essential features⁷ for efficient molecular recognition with our profen-based pentafluorophenyl active esters.⁸ We first studied the resolution of pentafluorophenyl 2-phenylbutanoate (rac)-5 using the enantiomerically pure 4,5-diphenylimidazolidine-2-thione (45,55)-4 (Scheme 2). Treatment of (4S,5S)-4 (1 equiv) with sodium hexamethyldisilazane (NaHMDS; 1.1 equiv) in THF at -78 °C, followed by the addition of pentafluorophenyl 2-phenylbutanoate (rac)-5 (1.1 equiv) gave, after stirring for 2 h, the corresponding 4,5-diphenylimidazolidine-2-thione (2R,4S,5S)-syn,anti-6 in 24% yield with 80% de (Scheme 2). The stereochemical outcome of this resolution was confirmed through stereospecific synthesis,⁹ and the diastereoselectivity was determined by 400 MHz ¹H NMR spectroscopy.¹⁰ Formation of this product, 4,5-diphenylimidazolidine-2-thione (2R,4S,5S)-syn,anti-6, is intrinsically low yielding (~25% yield)¹¹ due to its being more acidic than its starting 4,5-diphenylimidazolidine-2-thione, (45,55)-4, and the inherent enantiomeric composition of the active ester (rac)-5.12 The remaining pentafluorophenyl 2-phenylbutanoate (*rac*)-**5** was recovered in 41% yield with 38% ee. 13

With this information in hand, we next investigated the complementary resolution of racemic 4,5-diphenylimidazolidine-2-thione (4*R*5,5*R*5)-**4** using enantiomerically pure pentafluorophenyl 2-phenylbutanoate (*R*)-**5**, formed in 78% yield by DCC coupling of (*R*)-2-phenylbutanoic acid with pentafluorophenol (Scheme 3). Deprotonation of racemic 4,5-diphenylimidazolidine-2-thione (4*R*5,5*R*5)-**4** (1 equiv) with NaHMDS (1.1 equiv) in THF at $-78 \degree$ C, followed by the addition of pentafluorophenyl 2-phenylbutanoate (*R*)-**5** (1.1 equiv) gave, after 2 h at $-78 \degree$ C, 4,5-diphenylimidazolidine-2-thione (2*R*,4*S*,5*S*)-*syn,anti*-**6** in 28% yield with 78% de. The resolved 4,5-diphenylimidazolidine-2-thione (4*R*,5*R*)-**5** was recovered



Scheme 1. Resolution of oxazolidine-2-thione (rac)-1 using active ester (S)-2.

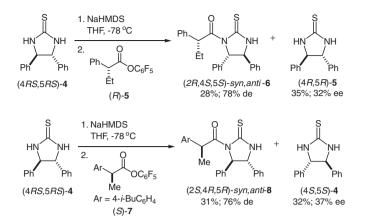


Scheme 2. Resolution of active ester (*rac*)-**5** using 4,5-diphenylimidazolidine-2-thione (45,55)-**4**.

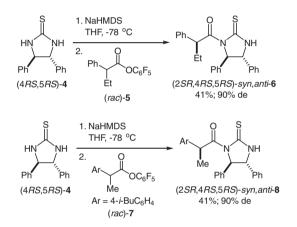


^{*} Corresponding author. Tel.: +44 1482 466401; fax: +44 1482 466410. *E-mail address:* j.eames@hull.ac.uk (J. Eames).

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Scheme 3. Kinetic resolution of 4,5-diphenylimidazolidine-2-thione (4*RS*,5*RS*)-**4** using active esters (*R*)-**5** and (*S*)-**7**.



Scheme 4. Mutual resolution of 4,5-diphenylimidazolidine-2-thione (4*RS*,5*RS*)-4 using active esters (*rac*)-5 and (*rac*)-7.

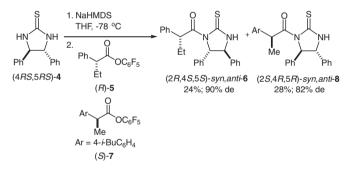
with the predicted stereochemistry in 35% yield with 32% ee (Scheme 3). The enantiomeric 4,5-diphenylimidazolidine-2-thione (4*S*,5*S*)-**4** was synthesised in 32% yield with 37% ee by treatment of the parent 4,5-diphenylimidazolidine-2-thione (4*R*,5*RS*)-**4** (1 equiv) with NaHMDS (1.1 equiv), followed by the addition of pentafluorophenyl 2-(4-isobutylphenyl)propanoate (*S*)-**7** (1.1 equiv) (Scheme 3). The complementary 4,5-diphenylimidazolidine-2-thione adduct (2*S*,4*R*,5*R*)-*syn*,*anti*-**8** was formed in 31% yield with 76% de (Scheme 3).

In an attempt to improve the efficiency of this resolution, we chose to resolve 4,5-diphenylimidazolidine-2-thione next (4RS,5RS)-4 in parallel^{2,14} using two complementary quasi-enantiomeric¹⁵ pentafluorophenyl active esters, (R)-5 and (S)-7. As a preliminary model study, we first investigated the mutual resolution of 4,5-diphenylimidazolidine-2-thione (4RS,5RS)-4, using the racemic active esters, (rac)-5 and (rac)-7 (Scheme 4). Deprotonaton of racemic 4,5-diphenylimidazolidine-2-thione (4RS,5RS)-4 (1 equiv) with NaHMDS (1.1 equiv) in THF at -78 °C, followed by addition of racemic active esters, (*rac*)-**5** (1.1 equiv) and (*rac*)-**7** (1.1 equiv). gave the 4,5-diphenylimidazolidine-2-thione adducts (2SR,4RS, 5RS)-syn,anti-6 (in 41% yield with 90% de) and (2SR,4RS,5RS)syn,anti-8 (in 41% yield with 90% de), respectively (Scheme 4). In both cases, the diastereoselectivity for these mutual resolutions was higher² than their corresponding kinetic resolutions; 90% de for both (rac)-5 and (rac)-7 versus 78% de for (R)-5 and 76% de for (*S*)-**7** (Schemes 3 and 4).

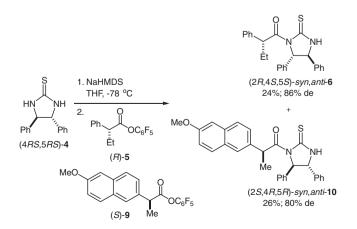
We next turned our attention to the parallel resolution of racemic 4,5-diphenylimidazolidine-2-thione (4*R*S,5*R*S)-**4** using a combination of *quasi*-enantiomeric active esters (*R*)-**5** and (*S*)-**7** (Scheme 5). Deprotonation of (4*R*S,5*R*S)-**4** (1 equiv) using NaHMDS (1.1 equiv) in THF at -78 °C followed by the addition of an equivmolar amount of *quasi*-enantiomeric active esters (*R*)-**5** (0.55 equiv) and (*S*)-**7** (0.55 equiv) gave, after 2 h, a mixture of two *quasi*-enantiomeric 4,5-diphenylimidazolidine-2-thiones (2*R*,4*S*,5*S*)-*syn,anti*-**6** and (2*S*,4*R*,5*R*)-*syn,anti*-**8** in 24% and 28% yields, with 90% de and 82% de, respectively (Scheme 5).

Increased separation of the 4,5-diphenylimidazolidine-2-thione adducts can be achieved using two *quasi*-enantiomeric active esters, (*R*)-**5** and (*S*)-**9**, which have different polarity.¹⁶ Treatment of 4,5-diphenylimidazolidine-2-thione (4*R*5,5*RS*)-**4** (1 equiv) with NaHMDS (1.1 equiv) in THF at -78 °C, followed by the addition of an equimolar amount of (*R*)-**5** (0.55 equiv) and (*S*)-**9**¹⁷ (0.55 equiv) gave the 4,5-diphenylimidazolidine-2-thiones (2*R*,4*S*,5*S*)-*syn,anti*-**6** and (2*S*,4*R*,5*R*)-*syn,anti*-**10**¹⁸ in 24% yield with 86% de and 26% yield with 80% de, respectively (Scheme 6). These adducts were easily separated by column chromatography; $\Delta R_{\rm F}$ [CH₂Cl₂] = 0.10.¹⁹

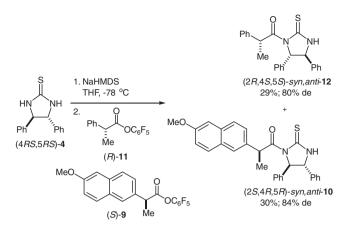
However, attempts at separating the enantiomers of (4RS,5RS)-4 using a pair of *quasi*-enantiomeric pentafluorophenyl propanoates, (R)-11²⁰ and (S)-9, gave the corresponding 4,5-diphenylimidazolidine-2-thiones (2R,4S,5S)-*syn,anti*-12²¹ and (2S,4R,5R)-*syn,anti*-10 in higher yields, 29% and 30%, respectively, but with decreased levels of diastereoselectivity, 80% de and 84% de, respectively (Scheme 7). These pentafluorophenyl propanoates, (R)-11 and (S)-9, were less diastereoselective than the pentafluorophenyl



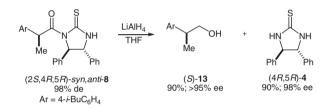
Scheme 5. Parallel resolution of 4,5-diphenylimidazolidine-2-thione (4*RS*,5*RS*)-**4** using active esters (*R*)-**5** and (*S*)-**7**.



Scheme 6. Parallel resolution of 4,5-diphenylimidazolidine-2-thione (4RS,5RS)-4 using active esters (*R*)-**5** and (*S*)-**9**.



Scheme 7. Parallel resolution of 4,5-diphenylimidazolidine-2-thione (4*RS*,5*RS*)-4 using active esters (*R*)-11 and (*S*)-9.



Scheme 8. LiAlH₄ reduction of 4,5-diphenylimidazolidine-2-thione (2S,4R,5R)-8.

butanoate (*R*)-**7**, due to their increased electrophilicity and less sterically demanding nature.

Access to the resolved parent 4,5-diphenylimidazolidine-2thione, such as (4*R*,5*R*)-**4**, was achieved by LiAlH₄ reduction of 4,5-diphenylimidazolidine-2-thione (2*S*,4*R*,5*R*)-*syn,anti*-**8** (Scheme 8). Addition of LiAlH₄ to a stirred solution of (2*S*,4*R*,5*R*)-*syn,anti*-**8** in THF at RT, and stirring the resulting solution for 2 h, gave the required enantiomerically pure 4,5-diphenylimidazolidine-2-thione (4*R*,5*R*)-**4**²² in 90% yield with 98% ee and the primary alcohol (*S*)-**13**^{5b,6} in 90% yield with >95% ee (Scheme 8).

In conclusion, we have shown²³ that four pentafluorophenyl active esters, (R)-5, (S)-7, (S)-9 and (R)-11, can be used to efficiently resolve racemic 4,5-diphenylimidazolidine-2-thione, (4RS,5RS)-4 in good yields. The levels of diastereocontrol were found to be good to excellent (78% de \rightarrow 90% de) favouring formation of the corresponding (2*R*^{*},4*S*^{*},5*S*^{*})-syn,anti-4,5-diphenylimidazolidine-2-thiones 6, 8, 10 and 12. The nearest analogy to this work is the desymmetrisation of meso-(4R,5S)-diphenylimidazolidine-2-thione²⁴ using profen-based pentafluorophenyl active esters. The levels of molecular recognition between the PhCHN(Na)C=S motif of meso-(4R,5S)-4,5-diphenylimidazolidine-2-thione and the enantiomerically pure active ester, (R)-5, were similarly high (92% de). In comparison, formation of the complementary diastereoisomeric anti-4,5-diphenylimidazolidine-2-thiones can be achieved using a deprotonation-methylation strategy developed by Davies²⁵ involving a related C2-symmetric N-propanoyl 4,5-diphenylimidazolidine-2-thione.

Acknowledgements

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- (a) Coulbeck, E.; Eames, J. Tetrahedron Lett. 2009, 50, 4449; (b) Coulbeck, E.; Eames, J. Synlett 2008, 333.
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- 7. For high levels of molecular recognition, the nucleophilic component must contain a PhCH(alkyl)-XM motif, where X is O and N.
- Addition of racemic sodiated *trans*-4,5-tetramethylene-imidazolidine-2-thione to (*S*)-pentafluorophenyl 2-(4-isobutylphenyl)propanoate leads to the corresponding (4*R*,5*R*)-3-[25-(4-isobutylphenyl)propanoyl]-4,5-tetramethyleneimidazolidine-2-thione in 36% yield with 40% de.
- 9. The 4,5-diphenylimidazolidine-2-thione (2R,4S,5S)-*syn,anti*-**6** was formed in 54% yield by the addition of lithiated 4,5-diphenylimidazolidine-2-thione (4S,5S)-**4** to (R)-2,4-dichlorophenyl 2-(4-isobutylphenyl)propanoate.
- 10. For 4,5-diphenylimidazolidine-2-thione (2*R*,4*S*,5*S*)-*syn*,*anti*-**6**, the PhCHNC=O doublet is at 5.37 ppm (1H, d, *J* = 5.4). Whereas, for (2*R*,4*R*,5*R*)-*anti*,*anti*-**6**, the PhCHNC=O doublet is at 5.22 ppm (1H, br d, *J* = 2.2).
- 11. Trace amounts (<2%) of dipropanonyl thioureas can be detected in the crude mixture by ¹H NMR spectroscopy.
- 12. Dipropanonyl thioureas can be formed using an excess of Brønsted base and active ester. Under our standard conditions, treatment of thiourea (45,55)-4 (1 equiv) with *n*-BuLi (2.2 equiv) and active ester (*rac*)-7 (4.4 equiv) gave the propanoyl thioureas *syn,anti* and *anti,anti*-8 in 25% yield in a diastereoisomeric ratio of 56:44 and the di-propanoyl thioureas *syn,anti,syn, syn,anti,anti* and *anti,anti* in 48% yield with a diastereoisomeric ratio of 35:45:20. The relative ratio of the propanoyl and di-propanoyl thioureas was 38:62 (by ¹H NMR spectroscopy).
- 13. The enantiomeric excess was determined by comparison with the known specific rotation, and through hydrolysis to give the corresponding carboxylic acid. The enantiomeric excess of this carboxylic acid was determined by selfcoupling to form the corresponding anhydride, see: Coulbeck, E.; Eames, J. Tetrahedron: Asymmetry 2009, 20, 635.
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- The 2-(6-methoxynaphthalen-2-yl)propanoyl transfer group, in (S)-9, is more polar than the 2-phenylbutanoyl transfer group in (R)-7.
- The mutual resolution using (rac)-9, gave (2RS,4SR,5SR)-syn,anti-10 in 31% yield with 72% de.
- 18. (2S,4R,5R)-syn,anti-**10**; $[\alpha]_D^{25}$ +166.8 (*c* 4.0, CHCl₃).
- (2R,4S,5S)-syn,anti-6: R_F[CH₂Cl₂] 0.66; (2S,4R,5R)-syn,anti-10; R_F[CH₂Cl₂] 0.56.
 The mutual resolution using (rac)-11, gave (2RS,4SR,5SR)-syn,anti-12 in 35%
- yield with 80% de.
- 21. (2R,4S,5S)-syn,anti-**12**; $[\alpha]_D^{25}$ -134.2 (*c* 2.4, CHCl₃).
- [a]²⁵_D +62.7 (c 2.2, CHCl₃) {lit. [a]²⁵_D +65.0 (c 0.25, CHCl₃); Davies, S. G.; Mortlock, A. A. Tetrahedron **1993**, 49, 4419).
- experimental procedure: Synthesis of (4S,5S)-3-[2R-23 Representative phenylbutanoyl]-4,5-diphenylimidazolidine-2-thione (2R,4S,5S)-syn,anti-6; NaHMDS (0.43 mL, 1.0 M in THF, 0.43 mmol) was added to a stirred solution of enantiomerically pure 4,5-diphenylimidazolidine-2-thione (45,55)-4 (0.1 g, 0.39 mmol) in THF at -78 °C. After stirring for 1 h, a solution of racemic pentafluorophenyl 2-phenylbutanoate (rac)-5 (0.14 g, 0.43 mmol) in THF (5 mL) was added. The resulting mixture was stirred for 2 h at -78 °C. The reaction was quenched with H₂O (10 mL). The organic layer was extracted with Et_2O (2 × 10 mL), dried (over MgSO₄) and evaporated under reduced pressure to give a separable mixture of two diastereoisomeric oxazolidine-2-ones (2R,4S,5S)-syn,anti-6 and (2S,4S,5S)-anti,anti-6 (ratio 90:10). The crude mixture was purified by flash column chromatography on silica gel eluting with light petroleum (bp 40-60 °C)/CH2Cl2 (1:9), to give (2R,4S,5S)-syn,anti-6 (33 mg, 21%) and (2S,4S,5S)-anti,anti-6 (4 mg, 3%) as a separable diastereoisomeric mixture (90:10-syn,anti-/anti,anti-); and pentafluorophenyl 2-phenylbutanoate (*R*)-**5** (26 mg, 41%) as a colourless oil with ~38% ee; $\{|\alpha|_D^{25}$ +26.4 (*c* 1.0, CHCl₃)); $\{(R)$ -; >98% ee $\{\alpha|_D^{25}$ +69.5 (*c* 5.3, CHCl₃)].

Characterisation data for: (2*S*,4*S*,5*S*)-*anti,anti*-6; colourless oil; $R_{\rm F}$ [CH₂Cl₂] 0.52; $v_{\rm max}$ (CHCl₃) cm⁻¹ 3022 (N–H), 1216 (C=S) and 1705 (C=O); $[\alpha]_D^{\rm 25}$ +7.4 (c 1.1, CHCl₃); $\delta_{\rm H}$ (400 MHz; CDCl₃) 7.42–7.38 (3 H, m, 3 × CH; Ph), 7.30–7.25 (5H, m, 5 × CH; Ph), 7.23–7.18 (5 H, m, 5 × CH; Ph), 6.84 (2 H, d, J 8.8, 2 × CH; Ph), 6.19 (1H, t, J 7.5, PhCHEt), 5.22 (1H, br d, J 2.2, PhCHNCO), 4.38 (1H, br s, PhCHN), 2.20–2.10 (1H, ddq, J 13.7, 7.5 and 7.3, CH_AH_BCH₃) and 0.74 (3H, t, J 7.3, CHCH₂CH₃); $\delta_{\rm C}$

 $\begin{array}{l} (100 \ \text{MHz}; \ \text{CDCl}_3) \ 180.4 \ (\text{C=S}), \ 175.1 \ (\text{C=O}), \ 140.5 \ (i-\text{C; Ph}), \ 139.2 \ (i-\text{C; Ph}), \ 139.2 \ (i-\text{C; Ph}), \ 139.2 \ (i-\text{C; Ph}), \ 129.3^2 \ (2C, \ 2 \times \text{CH}; \ \text{Ph}), \ 129.3^2 \ (2C, \ 2 \times \text{CH}; \ \text{Ar}), \ 128.8^2 \ (2 \times \text{CH}; \ \text{Ph}), \ 128.4^1 \ (1 \times \text{CH}; \ \text{Ph}), \ 128.3^2 \ (2C, \ 2 \times \text{CH}; \ \text{Ar}), \ 128.8^2 \ (2 \times \text{CH}; \ \text{Ph}), \ 128.4^1 \ (1 \times \text{CH}; \ \text{Ph}), \ 128.3^2 \ (2C, \ 2 \times \text{CH}; \ \text{Ar}), \ 127.1^1 \ (1 \times \text{CH}; \ \text{Ph}), \ 125.4^2 \ (2C, \ 2 \times \text{CH}; \ \text{Ph}), \ 127.1^1 \ (1 \times \text{CH}; \ \text{Ph}), \ 125.4^2 \ (2C, \ 2 \times \text{CH}; \ \text{Ar}), \ 127.1^1 \ (1 \times \text{CH}; \ \text{Ph}), \ 125.4^2 \ (2C, \ 2 \times \text{CH}; \ \text{Ar}), \ 127.1^1 \ (1 \times \text{CH}; \ \text{Ph}), \ 125.4^2 \ (2C, \ 2 \times \text{CH}; \ \text{Ar}), \ 127.1^1 \ (1 \times \text{CH}; \ \text{Ph}), \ 125.4^2 \ (2C, \ 2 \times \text{CH}; \ \text{Ar}), \ 127.1^1 \ (1 \times \text{CH}; \ \text{Ph}), \ 125.4^2 \ (2C, \ 2 \times \text{CH}; \ \text{Ar}), \ 127.1^1 \ (1 \times \text{CH}; \ \text{Ph}), \ 125.4^2 \ (2C, \ 2 \times \text{CH}; \ \text{Ar}), \ 127.1^1 \ (1 \times \text{CH}; \ \text{Ph}), \ 125.4^2 \ (2C, \ 2 \times \text{CH}; \ \text{Ar}), \ 127.1^1 \ (1 \times \text{CH}; \ \text{Ph}), \ 125.4^2 \ (2C, \ 2 \times \text{CH}; \ \text{Ar}), \ 127.1^1 \ (1 \times \text{CH}; \ \text{Ph}), \ 125.4^2 \ (2C, \ 2 \times \text{CH}; \ \text{Ar}), \ 127.1^1 \ (1 \times \text{CH}; \ \text{Ph}), \ 125.4^2 \ (2C, \ 2 \times \text{CH}; \ \text{Ar}), \ 127.1^1 \ (1 \times \text{CH}; \ \text{Ph}), \ 125.4^2 \ (2R, 45.55) \ \text{-syn}, \ \text{anti-e}; \ \text{colourless oil}; \ R_{\rm F} \ [\text{CH}_2\ (2L_2) \ 0.44; \ \nu_{\rm max} \ (\text{CHC}_3) \ \text{cm}^{-1} \ 3020 \ (\text{N-H}), \ 125.5 \ (\text{C=S}) \ \text{and} \ 1705 \ (\text{CH}; \ \text{Ph}), \ 123.4^2 \ (1 \times \text{H}, \ 175.4^2 \ \text{CH}), \ 123.4^2 \ (1 \times \text{H}, \ 175.4^2 \ \text{CH}), \ 123.4^2 \ (1 \times \text{H}, \ 175.4^2 \ \text{CH}), \ 123.4^2 \ (1 \times \text{H}, \ 175.4^2 \ \text{CH}), \ 123.4^2 \ (1 \times \text{H}, \ 175.4^2 \ \text{CH}), \ 123.4^2 \ (1 \times \text{H}, \ 175.4^2 \ \text{CH}), \ 123.4^2 \ (1 \times \text{H}, \ 175.4^2 \ \text{CH}), \ 123.4^2 \ (1 \times \text{H}, \ 175.4^2 \ \text{CH}), \ 123.4^2 \ (1 \times \text{H}, \ 175.4^2 \ \text{CH}), \ 123.4^2 \ (1 \times 175.4^2 \ \text{CH}), \ 123.4^2$

7.5 and 7.3, $CH_{A}H_{B}CH_{3}$) and 0.79 (3H, t, J 7.3, $CHCH_{2}CH_{3}$); δ_{C} (100 MHz; $CDCl_{3}$) 180.4 (C=S), 175.2 (C=O), 138.9 (i-C; Ph), 138.7 (i-C; Ph), 138.4 (i-C; Ar), 129.3^2 (2C, 2 \times CH; Ph), 129.0^3 (3C, 3 \times CH; Ph), 128.8^2 (2C, 2 \times CH; Ph), 128.2^2 (2C, 2 \times CH; Ph), 128.1^1 (1 \times CH; Ph), 126.9^1 (1 \times CH; Ar), 125.9^2 (2C, 2 \times CH; Ph), 125.7^2 (2C, 2 \times CH; Ar), 71.6 (PhCHNCO), 64.9 (PhCHN), 51.2 (PhCHEt), 26.8 (CH₂CH₃) and 12.0 (CH₂CH₃) (found MH^{*}, 401.1678; C₂₅H₂₅O₁N₂S₁⁺ requires MH^{*}, 401.1682).

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